This article was downloaded by: [University of Haifa Library]

On: 16 August 2012, At: 09:04 Publisher: Taylor & Francis

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH,

UK



Molecular Crystals and Liquid Crystals Science and Technology. Section A. Molecular Crystals and Liquid Crystals

Publication details, including instructions for authors and subscription information: http://www.tandfonline.com/loi/gmcl19

Electronic Structure of TiSe₂ and TiSe₂ Intercalated with Eu

S. Danzenbächer $^{\rm a}$, S. L. Molodtsova $^{\rm a}$, K. Koepernik $^{\rm b}$, Y. Tomm $^{\rm c}$ & C. Laubschat $^{\rm a}$

^a Inst. f. Oberflächen-und Mikrostrukturphysik, TU Dresden, D-01062, Dresden, Germany

b Max-Planck-Inst. f. Chemische Physik Fester Stoffe, D-01062, Dresden, Germany

^c Hahn-Meitner-Inst., D-14109, Berlin, Germany

Version of record first published: 27 Oct 2006

To cite this article: S. Danzenbächer, S. L. Molodtsova, K. Koepernik, Y. Tomm & C. Laubschat (2000): Electronic Structure of $TiSe_2$ and $TiSe_2$ Intercalated with Eu, Molecular Crystals and Liquid Crystals Science and Technology. Section A. Molecular Crystals and Liquid Crystals, 341:2, 45-50

To link to this article: http://dx.doi.org/10.1080/10587250008026115

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: http://www.tandfonline.com/page/terms-and-conditions

This article may be used for research, teaching, and private study purposes. Any substantial or systematic reproduction, redistribution, reselling, loan, sub-licensing, systematic supply, or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae, and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand, or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

Electronic Structure of TiSe₂ and TiSe₂ Intercalated with Eu

S. DANZENBÄCHER^a, S.L. MOLODTSOV^a, K. KOEPERNIK^b, Y. TOMM^c and C. LAUBSCHAT^a

^aInst. f. Oberflächen- und Mikrostrukturphysik, TU Dresden, D-01062 Dresden, Germany, ^bMax-Planck-Inst. f. Chemische Physik Fester Stoffe, D-01062 Dresden, Germany and ^cHahn-Meitner-Inst., D-14109 Berlin, Germany

We report on a PE study of the electronic structure of pristine TiSe₂. The experimental data are compared with results of band-structure calculations. It is confirmed that TiSe₂ reveals semimetallic properties. For the first time rare-earth metals were attempted to be incorporated in situ into TiSe₂. It is found that divalent Eu can be easy intercalated, whereas deposition of trivalent Gd leads to strong interfacial reactions. The electronic structure of the Eu-intercalation compound cannot be simply described within a rigid-band approach assuming charge transfer from Eu to unoccupied Ti d states. In a more elaborated approach hybridization phenomena have to be taken into consideration.

Keywords: photoemission; intercalation and rare-earth compounds

There has currently been considerable interest to transition-metal dichalcogenides (TMDC's) and their intercalation compounds (IC's) [11.2]. TMDC's consist of covalently bound sandwiches of the type DC-TM-DC, which are combined together by only weak van der Waals interaction. These quasi-two dimensional (2D) compounds serve as model systems to study the electronic structure of low-dimensional solids, transitions from 3D to 2D properties of electron states, and low-energy collective excitations like charge-density waves, which were observed in a number of TMDC's (TaS2, TaSe2, TiSe2, NbSe2) [2.3]. Detailed studies of the electronic structure of some TMDC's (e.g., dichalcogenides of Ti exploited as intercalation batteries and solar cells) lead, however, to inconsistent results. While TiTe2 was shown to be a semimetal, TiS2 is

assumed to be an intrinsic semiconductor ^[4]. For TiSe₂ some band structure calculations conclude metallic behavior with a considerable overlap between Ti d and Se p orbitals, while other calculations tend to prove a narrow-gap semiconducting behavior. Photoemission (PE) results have been interpreted as indicating either semimetallic ^[4] or semiconducting properties ^[5].

TMDC's are known to intercalate species into the van der Waals gap $^{[1]}$. Apart from conceivable applications as anisotropic conductors or highly ordered magnetic systems, the obtained IC's are of high importance to study basic phenomena relating to, e.g., an interplay between ionic and covalent chemical bondings upon intercalation. Quasi-2D structures can be tentatively used as rigid matrixes to fix guest atoms at certain distances from each other in order to follow a transition from bandlike to localized properties of their electronic states that is in particularly relevant for 4f and 5f elements $^{[6]}$. The intercalation may be carried out by exposing the samples at elevated temperatures to metal vapors in closed ampoules. Spontanious intercalation of alkalies takes place if the species are deposited on top of TMDC's in UHV conditions $^{[7]}$.

We report on a PE study of the electronic structure of TiSe₂. The data are compared with results of a full-potential nonorthogonal local-orbital minimumbasis band-structure calculations ^[8]. It is confirmed that TiSe₂ reveals semimetallic properties with electronic bands crossing the Fermi level (E_F) at the $\overline{\Gamma}$ and \overline{M} points in the Brillouin zone (BZ). For the first time rare-earth's (RE's) were attempted to be incorporated *in situ* into TiSe₂. It is found that similar to alkalies divalent Eu can be easy intercalated into the TMDC, whereas deposition of trivalent Gd leads to a strong interfacial reaction between Gd and Se. The electronic structure of the Eu-IC cannot be interpreted simply in terms of a rigid-band approach assuming charge transfer from Eu to Ti d states. In a more elaborated model hybridization effects have to be taken into account.

The measurements were performed at the Berliner Elektronenspeicherring für Synchrotronstrahlung (BESSYI) using radiation from the TGM4 and SX/700 II beamlines. Valence-band and core-level PE spectra were taken with a hemispherical electron energy analyzer (ARIES-VSW) tuned to an energy re-

solution of 150 meV and an angle resolution of 1°. Similar to our studies of RE-graphite IC's ^[6,9], for the *in-situ* intercalation of Eu and Gd a method of thermal deposition of thick layers of intercalant (100 Å) onto the TMDC surface followed by a step-by-step annealing was exploited. In both cases the deposition at room temperature resulted in nonordered interfaces. Several stages of annealing (up to 700°C) of the systems led in case of Eu/TiSe₂ to a recovering of a crystalline structure with a sharp TMDC-like hexagonal LEED pattern.

Energy distribution curves (EDC's) taken along the $\overline{\Gamma}$ - \overline{K} direction in the

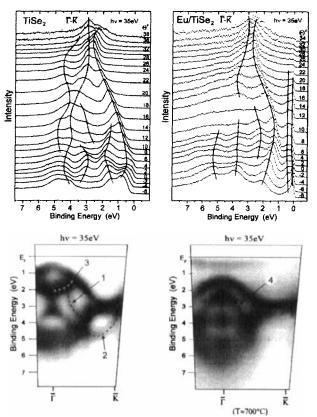


FIGURE 1 PE response of TiSe₂ and Eu-IC in form of EDC's and 2D plots visualizing intensity along the $\overline{\Gamma}$ - \overline{K} direction.

surface BZ of TiSe₂ are shown in Fig. 1. The experimental data measured along this and other high symmetry directions are compared with the calculated bulk band structure projected onto the surface of the TMDC (Fig. 2). We emphasize a good correspondence between experimental and theoretical results particularly in the region of the \overline{K} and \overline{M} points, where both measured and theoretical bands reveal predominantly 2D behavior. In the region of the $\overline{\Gamma}$ point electronic bands are found to be clearly three dimensional. There are only two places in the BZ: the $\overline{\Gamma}$ and \overline{M} points, with nonzero density of states at E_F . In both cases the semimetallic properties arise from hybridization between Se 4*p* [highly dispersive upper valence bands (VB's)] and Ti 3*d* (flat bands

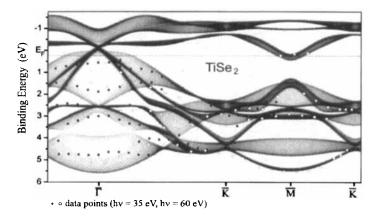


FIGURE 2 Projected band structure (calculated E_F is marked by a light horizontal line) in comparison to the experiment data points.

at the conduction band bottom) orbitals originating from neighboring layers.

EDC's measured from the Eu/TiSe₂ system after annealing at 700°C are shown in the right top panel of Fig. 1. At first glance the two sets of PE spectra in Fig. 1 replicate each other with the main difference that the EDC's presented in the right panel are shifted in average by about 0.8 eV toward higher binding energies (BE's) as expected for intercalated TMDC ^[10]. Additional structures

appearing at E_F at any angle of analyzing can be assigned to the flat Ti $3d_z^2$ band, which is filled in the IC due to a charge transfer from Eu atoms.

A simple rigid-band model, however, is not appropriate for the Eu-IC. As seen from the 2D representations visualizing the PE intensities of TiSe₂ and the IC along the $\overline{\Gamma}$ - \overline{K} direction in the surface BZ of TiSe₂ (Fig. 1) there are a number of TiSe₂ bands (marked by numbers), which have no clear counterparts in the Eu-IC and *vise versa*. The latter points to a change of chemical interaction in the IC caused by hybridization between chemically active 5d states of Eu and 4p states of Se. Chemical interactions are not that dramatic for the divalent RE with only slight admixture of the d character to the VB as may be concluded from the Se 3d core-level spectrum that do not show chemical

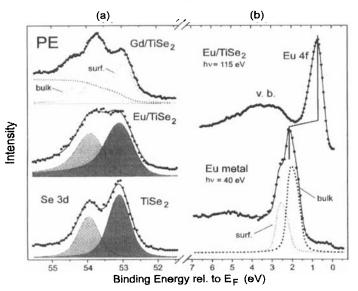


FIGURE 3 (a) Se 3d PE spectra. Shadowed subspectra simulate contributions from $3d_{3/2}$ (left) and $3d_{5/2}$ (right) components; (b) Eu 4f PE spectra.

shifts upon intercalation [Fig. 3(a)]. The difference in the PE branching ratio for the two spectra measured for TiSe₂ and Eu-IC can be explained by an in-

crease of line asymmetry due to electron-hole excitations in the metallic compound. A least-squares fit of the data applying Doniach - Sunjic line-shapes [11] with different asymmetry parameters shows that both BE and branching ratio of the Se 3d emission are not changed in the compound.

Much stronger chemical interactions were observed upon deposition of trivalent Gd onto TiSe₂ followed by annealing at 700°C. As seen in Fig. 3(a) in contrast to the Eu/TiSe₂ system the Se 3d spectrum of the Gd compound reveals a triplet structure, which can be simulated ^[12] assuming two components assigned to bulk and surface contributions from Gd selenide. The chemical interaction destroys the TiSe₂ matrix resulting in a nonordered interface as it is also concluded from the LEED experiment.

As follows from analysis of the 4f PE spectra [Fig. 3(b)], Eu remains divalent in the IC. A shift of the 4f signal toward E_F in the IC as compared to its position in Eu metal, which is similar to that observed in Eu-graphite IC ^[9], can be understood by the increase of the cohesive energy upon intercalation.

Acknowledgments

This work was supported by the Graduiertenkolleg "Struktur und Korrelationseffekte in Festkörpern" and the Sonderforschungsbereich 463, TP B4.

References

- [1] Intercalation in Layered Materials, NATO ASI Series, Series B: Physics Vol. 148, ed by M.S. Dresselhaus (Plenum Press, NY and London, 1986).
- [2] R.H. Friend and A.D. Yoffe, Adv. Phys., 36, 1 (1987).
- [3] A. Kikuchi and M. Tsukada, Surf. Sci., 326, 195 (1995).
- [4] O. Andersen et al., Phys. Rev. Lett., 55, 2188 (1985).
- [5] N.G. Stoffel et al., Phys. Rev. B, 31, 8049 (1985).
- [6] S. Danzenbächer et al., Physica B, 259-261, 1153 (1999).
- [7] H.I. Starnberg and H.P. Hughes, J. Phys. C 20, L97 (1987); A. Schellenberger et al. Phys. Rev. B, 45, 3538 (1992).
- [8] K. Koepernik and H. Eschrig, Phys. Rev. B, 59, 1743 (1999).
- [9] S.L. Molodtsov et al., Phys. Rev. B, 53, 16621 (1996).
- [10] C. Umrigar et al., Phys. Rev. B, 26, 4953 (1982).
- [11] S. Doniach and M. Sunjic, J. Phys. C, 3, 285 (1970).
- [12] G. Kaindl et al., Phys. Rev. B, 26, 1713 (1982).